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Key indicators

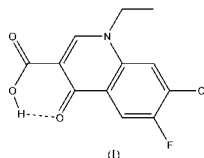
Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.036
 wR factor = 0.104
Data-to-parameter ratio = 13.8For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.7-Chloro-1-ethyl-6-fluoro-1,4-dihydro-4-oxo-
quinoline-3-carboxylic acid

The title compound, $\text{C}_{12}\text{H}_9\text{ClFNO}_3$, was synthesized from ethyl 2,4-dichloro-5-fluorobenzoylacetate. The quinoline ring has a same plane with the carboxyl group.

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Comment

Norfloxacin [1-ethyl-6-fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)-3-quinolinecarboxylic acid] is a fluorinated quinolone antibacterial agent (Koga *et al.*, 1980). Fluorinated quinolone compounds are characterized by having an F atom at the 6-position and a substituted amine group at the 7-position. The title compound, (I), is an intermediate in the synthesis of norfloxacin; the molecular structure of (I) is illustrated in Fig. 1. The quinoline ring has a same plane with the carboxyl group. In the crystal structure of (I), molecules are connected into two-dimensional layers, where are existing $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{Cl}$ hydrogen bonds.



Experimental

The title compound was prepared according to the method of Irikura (1978) from ethyl 2,4-dichloro-5-fluorobenzoylacetate. Condensation of ethyl 2,4-dichloro-5-fluorobenzoylacetate with triethyl orthoformate by refluxing in acetic anhydride produced ethyl 2-(2,4-dichloro-5-fluorobenzoyl)-3-ethoxyacrylate. The intermediate was reacted without further purification with ethanamine in methylene chloride to afford ethyl 2-(2,4-dichloro-5-fluorobenzoyl)-3-ethan-aminoacrylate. This was cyclized by heating with sodium hydride to give 7-chloro-1-ethyl-6-fluoro-1,4-dihydro-3-ethoxycarbonyl-4-oxoquinoline. The ester was hydrolyzed by heating with aqueous KOH in tetrahydrofuran to give 7-chloro-1-ethyl-6-fluoro-1,4-dihydro-4-oxoquinoline-3-carboxylic acid. The crude product was purified by

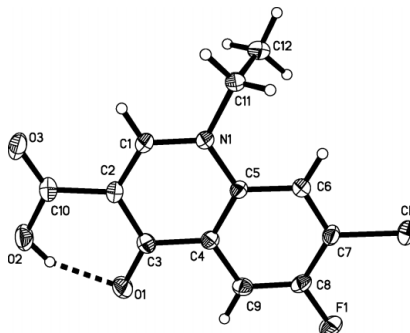


Figure 1

The molecular structure of (I), drawn with 30% probability ellipsoids. The intramolecular hydrogen bond is indicated by a dashed line.

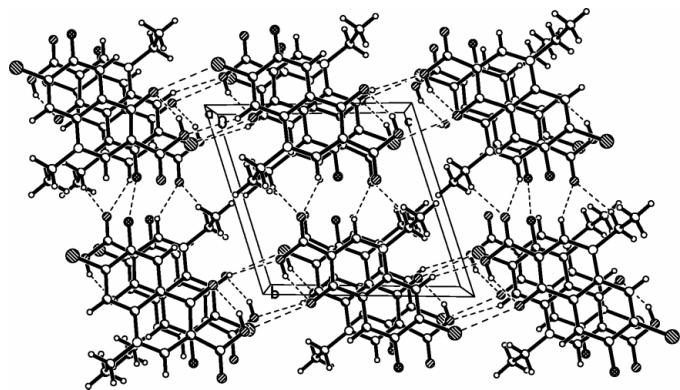


Figure 2
The crystal structure of (I), viewed along the *a* axis.

recrystallization from *N,N*-dimethylformamide. The product (20 mg) was dissolved in dichloromethane (20 ml) and the solution kept at room temperature for 6 d. Slow evaporation produced colorless single crystals of (I) suitable for X-ray analysis.

Crystal data

$C_{12}H_9ClFNO_3$	$Z = 2$
$M_r = 269.65$	$D_x = 1.603 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 7.145 (3) \text{ \AA}$	Cell parameters from 792 reflections
$b = 8.915 (4) \text{ \AA}$	$\theta = 3.7\text{--}26.4^\circ$
$c = 9.375 (4) \text{ \AA}$	$\mu = 0.35 \text{ mm}^{-1}$
$\alpha = 71.896 (6)^\circ$	$T = 293 (2) \text{ K}$
$\beta = 80.025 (7)^\circ$	Block, colorless
$\gamma = 85.109 (7)^\circ$	$0.36 \times 0.34 \times 0.28 \text{ mm}$
$V = 558.7 (4) \text{ \AA}^3$	

Data collection

Bruker SMART CCD area-detector diffractometer	2273 independent reflections
φ and ω scans	1745 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Bruker, 1997)	$R_{\text{int}} = 0.015$
$T_{\text{min}} = 0.809$, $T_{\text{max}} = 0.906$	$\theta_{\text{max}} = 26.5^\circ$
3221 measured reflections	$h = -6 \rightarrow 8$
	$k = -11 \rightarrow 10$
	$l = -11 \rightarrow 11$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.104$
 $S = 1.06$
 2273 reflections
 165 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0584P)^2 + 0.0322P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.19 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.29 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bonding geometry (\AA , $^\circ$).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
$O2\text{--}H2\cdots CH^i$	0.82	2.98	3.546 (2)	129
$O2\text{--}H2\cdots O1$	0.82	1.79	2.550 (2)	155

Symmetry code: (i) $x, y, 1 + z$.

H atoms were positioned geometrically and refined with $C\text{--}H = 0.93\text{--}0.98 \text{ \AA}$ and refined in a riding model with $U_{\text{iso}}(H) = 1.2 U_{\text{eq}}(\text{carrier})$.

Data collection: SMART (Bruker, 1997); cell refinement: SMART; data reduction: SAINT (Bruker, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

References

- Bruker (1997). *SADABS*, *SMART*, *SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Irikura, T. (1978). US Patent No. 4 14 719.
- Koga, H., Itoh, A., Murayana, S. & Suzue, S. (1980). *J. Med. Chem.* **23**, 1358–1363.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.